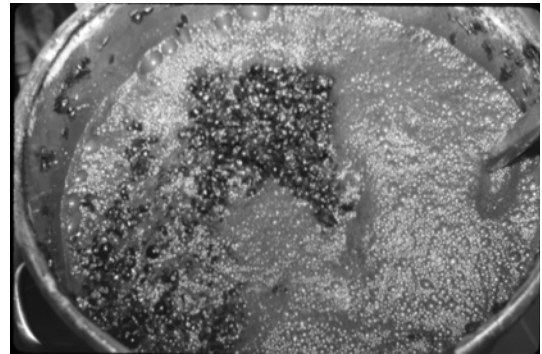


Vineyard Site  
 Cultivar Selection  
 Vineyard Management  
 Grape Maturity & Harvest  
 Juice Extraction  
**Alcoholic Fermentation**  
 Malolactic Fermentation  
 Wine Aging (Storage)

Cornell University

Alcoholic Fermentation - fermentation pathway



Cornell University

**Alcoholic Fermentation** (acetaldehyde final electron acceptor)  
 $C_6H_{12}O_6 \rightarrow 2 C_2H_5OH + 2 CO_2 + 2 ATP + \text{heat}$   
 glucose ethanol

**Respiration** (complete oxidation)  
 $C_6H_{12}O_6 + 6 O_2 \rightarrow 6 CO_2 + 6 H_2O + 36 ATP + \text{heat}$

Cornell University

Yeast cell - membrane structures

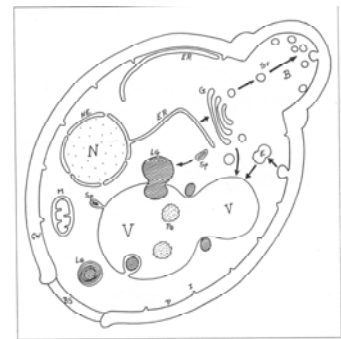


Figure 8 : Yeast molecular structure.  
 C= Cell wall - B= Bud scar - P= Plasmalemma - I= Plasma membrane invagination - V= Vacuole - PG= Phosphate granule - LG= Lipid granule - Sp= Sporosome - M= Mitochondria - N= Nucleus - NE= Nuclear envelope - ER= Endoplasmic reticulum - G= Golgi - E= Endosome - B= Bud - Sv= Secretory vesicles.

Cornell University

Yeast cell wall

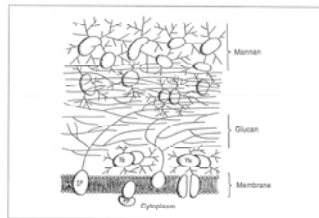
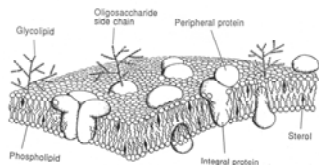


Figure 7 Cell Wall Structure according to Sakubandhu and Neufeld. *J Cell Growth Mech Biol* 1989; 1: 103-110.  
 C= Chitin - M= Mannan - PE= Peptidoglycan - PM= Plasma membrane - IB= Integral membrane protein - PP= Peripheral membrane protein.

Yeast cell membrane



Cornell University

Alcoholic Fermentation

Uptake of glucose:  
 Glucose  $\rightarrow$  glucose-6P

Glucose-6phosphate to fructose-1,6 diphosphate

FIRST STAGE -2 ATP

Cornell University

### Alcoholic Fermentation

fructose-1,6-diphosphate  $\rightarrow$  dihydroxyacetone-P + glyceraldehyde-3-P  
aldolase

dihydroxyacetone-P  $\rightarrow$  glyceraldehyde-3-P  
triosephosphate isomerase

glyceraldehyde-3-P + NAD + Pi  $\rightarrow$  1,3-diphosphoglyceric acid + NADH+H<sup>+</sup>  
triosephosphate dehydrogenase

NET GAIN: + 2 NADH+H<sup>+</sup>

Cornell University

### Alcoholic Fermentation

1,3-diphosphoglyceric acid + ADP  $\rightarrow$  3-phosphoglyceric acid + ATP  
3-phosphoglycerate kinase

3-phosphoglyceric acid  $\rightarrow$  2-phosphoglyceric acid  
phosphoglyceromutase

2-phosphoglyceric acid  $\rightarrow$  PEP + H<sub>2</sub>O  
enolase

PEP + ADP  $\rightarrow$  pyruvic acid + ATP  
pyruvate kinase

CH<sub>3</sub>-CO-COOH    CH<sub>3</sub>-COH    CH<sub>3</sub>-CH<sub>2</sub>OH  
pyruvic acid  $\rightarrow$  acetaldehyde + CO<sub>2</sub> + NADH<sub>2</sub>  $\rightarrow$  ethanol + NAD  
pyruvate decarboxylase    alcohol dehydrogenase

SECOND STAGE  
Cornell University

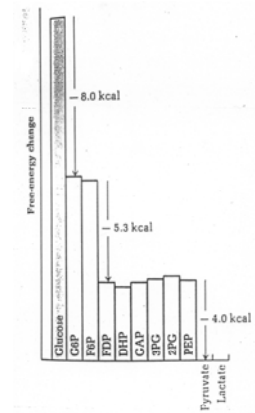
- 2 NADH+H<sup>+</sup>  
+ 4 ATP  
NET                    2 ATP

### Alcoholic Fermentation - energy balance

FIRST STAGE	- 2 ATP
	+ 2 NADH+H <sup>+</sup>
	- 2 NADH+H <sup>+</sup>
SECOND STAGE	+ 4 ATP
NET	2 ATP

Cornell University

### Energy barriers



Cornell University

### Alcoholic Fermentation

C<sub>6</sub>H<sub>12</sub>O<sub>6</sub>  $\rightarrow$  2 C<sub>2</sub>H<sub>5</sub>OH + 2 CO<sub>2</sub> + 2 ATP + heat  
hexose                    ethanol                    gas                    energy

Total free energy change: -227 kcal  
Conserved energy: 15 kcal (6.6%)

### (aerobic) Respiration

C<sub>6</sub>H<sub>12</sub>O<sub>6</sub> + 6O<sub>2</sub>  $\rightarrow$  6H<sub>2</sub>O + 6CO<sub>2</sub> + 36 ATP + heat  
hexose                    oxygen                    water                    gas                    energy

Total free energy change: -686 kcal  
Conserved energy: 270 kcal (40%)

Cornell University

### Alcoholic Fermentation - History

C<sub>6</sub>H<sub>12</sub>O<sub>6</sub> + 2 Pi  $\rightarrow$  2 C<sub>2</sub>H<sub>5</sub>OH + 2 CO<sub>2</sub> + 2 ATP + heat

#### yeast - cause of fermentation

1632-1723 Antonie van Leeuwenhoek  
1789 Lavoisier: fermentation end products ethanol and CO<sub>2</sub>  
1810 Gay-Lussac: C<sub>6</sub>H<sub>12</sub>O<sub>6</sub>  $\rightarrow$  2 C<sub>2</sub>H<sub>5</sub>OH + 2 CO<sub>2</sub>  
1818 Erxleben, De La Tour, Schwann, Kuetzing (1837): yeasts are the cause  
1866 Louis Pasteur: études sur le vin, 1876 études sur la bière  
Christian Emil Hansen, Hermann Müller-Thurgau, Julius Wortmann (1894)  
pure culture yeast

Cornell University

Biochemistry of Alcoholic Fermentation

1897 Buchner: extract of yeast retains ability to ferment glucose to ethanol
1905 Harden and Young: heat-labile fraction (zymase, proteins) + heat-stable fraction (cozymase, NAD, ATP, ADP)
Gustav Embden: cleavage of fructose 1,6 -di-P
Otto Mayerhof: verified Embden's theory and added energetics
1940: Embden-Mayerhof pathway: glucose --> pyruvate
also: Otto Warburg, CF Cori and GT Cori, J Parnas

Cornell University

Alcoholic Fermentation - Uptake of Sugars

Passage through cell wall and cell membrane

3 basic entry mechanisms for uptake of solutes (nutrients) across cell membrane:
simple diffusion, facilitated (carrier mediated) diffusion, active transport

in Saccharomyces crevisiae, glucose, fructose, and mannose are transported via facilitated diffusion, a non-concentrative process, several carriers

sucrose is cleaved outside the cell --> glucose + fructose
fructofuranidase (=saccharase, =invertase)
yeast has an extracellular and intracellular invertase located on the outside of the yeast cell wall

glu and fru transported into the cell

Cornell University

Alcoholic Fermentation - Regulation of Pathway

MULTISTEP REGULATION OF SUGAR FERMENTATION PATHWAY

- uptake and phosphorylation (hexokinase)
multienzyme complex of glycolytic enzymes

Cornell University

Alcohol Yield

Table with 3 columns: C6H12O6, 2 C2H5OH, 2 CO2. Rows show molar and gram amounts and percentages.

actual observed yield: 47%
220 g/L sugar x 0.47 ----> 103.4 g/L = 10.3% ethanol

why only 47%?
loss of ethanol use of C for building cells production of other products (flavors!)

Cornell University

Alcoholic Fermentation Byproducts - CO2

The volume of CO2 produced is about 40 to 50 times larger than that of the fermenting must.
1 Mol of CO2 is 22.4 L (180 g/L sugar (1 Mol) ----> 44.8 L)

CO2 is poisonous!!
3-4% breathing difficulties
10% severe shortage of breath, a burning candle extinguishes at 10% CO2
15% fainting, death

Fermentation rooms must be well ventilated!

Cornell University

Alcoholic Fermentation Byproducts - Heat

C6H12O6 ----> 2 C2H5OH + 2 CO2 + 2 H2O + 2 ATP
1 Mol glucose (180 g) ----> 24 kcal
must with 180 g/L sugar ----> 24 kcal
220 g/L sugar ---->
220 x 24 = 29 kcal/L
180

Possible temp. increase: initial must temp, 16°C + 29 = 45°C!

Approx. 20% of the heat prod. are carried away by the escaping CO2

Cornell University

*Alcoholic Fermentation Byproducts - Yeast Biomass*

0.8 - 1.3% of C of sugar is incorporated into yeast biomass during wine fermentation

yeast lees (incl. 2/3 grape particles): 1.5 - 3% of ferm. volume

**Cornell**  
University

*Metabolic Byproducts of Alcoholic Fermentation*

**PRIMARY BYPRODUCTS**

pyruvic acid, acetaldehyde, ketoglutaric acid (binding partners for SO<sub>2</sub>!)

glycerol (4-14g/L)

dihydroxyacetone-P + NADH<sub>2</sub> → a-glycero-P + NAD → glycerol + Pi

lactic acid (0.1-0.2 g/L) (lack of thiamin to decarboxylate pyruvate to acetaldehyde)

CH<sub>3</sub>COCOOH → CH<sub>3</sub>CHOHCOOH

pyruvic acid + NADH<sub>2</sub> → lactic acid + NAD

acetic acid (0.2-0.5 g/L):

CH<sub>3</sub>COH + H<sub>2</sub>O + NAD → CH<sub>3</sub>COOH + NADH<sub>2</sub>

oxidation of acetaldehyde to acetic acid

**Cornell**  
University

*Metabolic Byproducts of Alcoholic Fermentation*

**SECONDARY BYPRODUCTS**

2,3-butanediol acetoin diacetyl

higher alcohols

these fermentation byproducts are part the cell building metabolism (byproducts of AA synthesis valine, leucine, isoleucine)

Sulfur compounds (SO<sub>2</sub>, H<sub>2</sub>S, mercaptans,...)

**NUTRIENT LIMITATIONS**

Nitrogen (amino acids, short peptides, ammonia)

Pi (H<sub>3</sub>PO<sub>4</sub>)

nicotinic acid (B-vitamin) important for NAD (nicotinamid adenine dinucleotide)

thiamin (B1) (pyruvate → acetaldehyde), lack: accumulation of pyruvate and ketoglutarate

**Cornell**  
University